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formula for an ionic  
compound represents one  
formula unit - the simplest  
ratio of the compound's

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positive ions (cations) and its negative ions (anions). Monatomic cations are identified simply by the element's name. Monatomic anions more complicated. The ending of the element's name is dropped, the, "ide" added

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atom. compounds composed of  
two elements. naming  
system;... name of the  
cation is given first,

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followed by... polyatomic ions  
that contain oxygen;... an  
anion that has one few...  
monatomic ions. ions formed  
from a single atom. binary  
compounds. compounds  
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Sarquis. Chapter 5 The

Periodic Law. Educators. HC

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Problem 1 Describe the

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contributions made by the following scientists to the development of the periodic table: a. Stanislao Cannizzaro b. Dmitri Mendeleev c. Henry Moseley.  
HC Hashim C. Numerade  
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ANSWER Answer the following  
questions in the space  
provided. 1. Label each of  
the following statements as

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7 Test Answers True or False: True a. If the formula mass of one molecule is  $x$  amu, the molar mass is  $x$  g/mol. False b.



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