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Unit 6 - Test(Chemical Compound) -
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The lines/curves indicate the maximum amount of solute which can be dissolved in 100g of water at each temperature. We call this a saturated solution.. Any amount above the curve is said to be supersaturated; there is more solute than the solvent can hold, therefore the excess precipitates out/settles to the bottom.. Any amount below the line is called unsaturated, meaning more solute

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can be ...

Unit 6 | chemistry

UNIT 6: RADIOACTIVITY AND

NUCLEAR CHEMISTRY 5. A skeleton of a

mammoth was analysed and found to

contain 21.2% of the carbon-14 found in

living bone. The half-life of carbon-14 is

5730 years. What is the most likely year of

the mammoth's death? Use $\log(N_t/N_0) = -\lambda t$

$\ln(0.212) = -\lambda t$, $t_{1/2} = 5730$

years; hence find t This year – t = year of

death

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UNIT 6 – REDOX REACTIONS 6 • The oxidation number of an atom is the charge that would exist on an individual atom if the bonding were completely ionic • In simple ions, the oxidation number of the atom is the charge on the ion: - Na^+ , K^+ , H^+ all have an oxidation number of +1 - Mg^{2+} , Ca^{2+} , Pb^{2+} all have an oxidation number of

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+2 - Cl-, Br-, I-all have an oxidation number of -1

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