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Chemical and  
Physical Changes**  
Change Chapter

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Pure Substances  
and Mixtures!  
(Classification  
of Matter)

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ozone formation,  
chlorofluorocarb  
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Section 2 MATTER  
and Its

Properties Matter

– Anything that  
has MASS and

VOLUME. ATOMS

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and MOLECULES

the basic building blocks of matter.

ATOM-- the smallest unit of an element that still maintains the properties that element.

ELEMENT-- a pure substance made up of only one type of atom.

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in Atoms

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the definition  
in Column A with  
the term in  
Column B. Column  
A Column B 1.

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frequencies of  
the  
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properties

(examples)

intensive

property

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property that

depends on the

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property that  
depends on the  
type of matter  
in a sample  
not... mass,  
volume, length,  
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H C H O O O CO

CO CO H O H O H

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0 H O C H O CO

use mole ratio 5

mol 0 2 1 mol C

3 H 8 3 mol CO 2

1 mol C 3 H 8 4

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C 3 H 8 1 mol C

3 H 8 5 mol 0 2

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of a chemical  
change.

Properties of  
Compounds Quite  
different  
properties than  
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