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chemical potential is the molar
Gibbs energy: $dG = (\mu_2 - \mu_1)dn$
so $\Delta G = (\mu_2 - \mu_1)n = (-8.3 \text{ kJ}$
 $\text{mol}^{-1})(0.15 \times 10^{-3} \text{ mol}) =$
 $+1.2 \times 10^{-3} \text{ kJ} = 1.2 \text{ J}.$

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of Chemistry, University of Oxford and fellow of Lincoln College, Oxford) and Julio De Paula (Professor and Dean of the College of Arts and Sciences, Lewis and Clark College, Portland, Oregon) and published by Oxford University Press in 2006.

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change in the number of atoms or
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