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Chapter 6 – Chemical Composition

6.chemical bonding

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Chapter 6 – Chemical Bonds

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See the world, one molecule at a time. Chemistry helps us understand not only the world around us, but also our own bodies. CHEMISTRY MADE SIMPLE makes it fun. Each chapter has practice problems with complete solutions that reinforce learning. A glossary of chemical terms, the modern periodic table, and detailed illustrations throughout make this the best introduction to one of the most studied of all sciences. Topics covered include: *the Scientific Method *the structure and properties of matter *compounds *laws of chemistry *gases, liquids, and solids *solutions *electrochemistry *the atmosphere *biochemistry *organic chemistry *nuclear chemistry *energy *the environment Look for these Made Simple titles Accounting Made Simple Arithmetic Made Simple Astronomy Made Simple Biology Made Simple Bookkeeping Made Simple Business Letters Made Simple Earth Science Made Simple English Made Simple French Made Simple German Made Simple Ingles Hecho Facil Investing Made Simple Italian Made Simple Latin Made Simple Learning English Made Simple Mathematics Made Simple The Perfect Business Plan Made Simple Philosophy Made Simple Physics Made Simple Psychology Made Simple Sign Language Made Simple Spelling Made Simple Statistics Made Simple Your Small Business Made Simple www.broadwaybooks.com

When you're cooking, you're a chemist! Every time you follow or modify a recipe, you are experimenting with acids and bases, emulsions and suspensions, gels and foams. In your kitchen you denature proteins, crystallize compounds, react enzymes with substrates, and nurture desired microbial life while suppressing harmful bacteria and fungi. And unlike in a laboratory, you can eat your experiments to verify your hypotheses. In Culinary Reactions, author Simon Quellen Field turns measuring cups, stovetop burners, and mixing bowls into graduated cylinders, Bunsen burners, and beakers. How does altering the ratio of flour, sugar, yeast, salt, butter, and water affect how high bread rises? Why is whipped cream made with nitrous oxide rather than the more common carbon dioxide? And why does Hollandaise sauce call for &"clarified&" butter? This easy-to-follow primer even includes recipes to demonstrate the concepts being discussed, including: &· Whipped Creamsicle Topping—a foam &· Cherry Dream Cheese—a protein gel &· Lemonade with Chameleon Eggs—an acid indicator

Ideas of Quantum Chemistry shows how quantum mechanics is applied to chemistry to give it a theoretical foundation. The structure of the book (a TREE-form) emphasizes the logical relationships between various topics, facts and methods. It shows the reader which parts of the text are needed for understanding specific aspects of the subject matter. Interspersed throughout the text are short biographies of key scientists and their contributions to the development of the field. Ideas of Quantum Chemistry has both textbook and reference work aspects. Like a textbook, the material is organized into digestible sections with each chapter following the same structure. It answers frequently asked questions and highlights the most important conclusions and the essential mathematical formulae in the text. In its reference aspects, it has a broader range than traditional quantum chemistry books and reviews virtually all of the pertinent literature. It is useful both for beginners as well as specialists in advanced topics of quantum chemistry. The book is supplemented by an appendix on the Internet. * Presents the widest range of quantum chemical problems covered in one book * Unique structure allows material to be tailored to the specific needs of the reader * Informal language facilitates the understanding of difficult topics

Informed by new planetary discoveries and the findings from recent robotic missions to Mars, Jupiter, and Saturn, scientists are rapidly replacing centuries of speculation about potential extraterrestrial habitats with real knowledge about the possibility of life outside our own biosphere -- if it exists, and where. This second edition of Kevin W. Plaxco and Michael Gross's widely acclaimed text incorporates the latest research in astrobiology to bring readers the most comprehensive, up-to-date, and engaging introduction to the field available. Plaxco and Gross expand their examination of the origin of chemical elements, the developments that made the Universe habitable, and how life continues to be sustained. They discuss in great detail the formation of the first galaxies and stars, the diverse chemistry of the primordial planet, the origins of metabolism, the evolution of complex organisms, and the feedback regulation of Earth's climate. They also explore life in

extreme habitats, potential extraterrestrial habitats, and the current status of the search for extraterrestrial life. Weaving together the relevant threads of astronomy, geology, chemistry, biophysics, and microbiology, this broadly accessible introductory text captures the excitement, controversy, and progress of the dynamic young field of astrobiology. New to this edition is a glossary of terms and an epilogue recapping the key unanswered questions, making Astrobiology an ideal primer for students and, indeed, for anyone curious about life and the Universe. Praise for the first edition of Astrobiology "Certainly the most readable introduction to astrobiology now available." -- Chemical and Engineering News "Plaxco and Gross bring us as close to aliens as we can currently get. I recommend this book to anyone interested in science's newest kid on the block." -- Astronomy Now "A good read for all those who are fascinated by the search for extraterrestrial life and the origin of life on our own planet. I shall certainly value it in my own library." -- Chemistry World "An accessible guide to this young and interdisciplinary field." -- Physics World "The fascinating world of extremophiles is well presented, and a broad overview of the searches for evidence of life beyond Earth rounds off the book. The text is liberally illustrated with relevant figures that greatly enhance the content, and entertaining snippets of information detailing the quirks of research in this field nicely supplement the scientific content." -- Astrobiology "A comprehensive yet concise introduction to the field." -- The Space Review

Every day, billions of photographs, news stories, songs, X-rays, TV shows, phone calls, and emails are being scattered around the world as sequences of zeroes and ones: bits. We can't escape this explosion of digital information and few of us want to-the benefits are too seductive. The technology has enabled unprecedented innovation, collaboration, entertainment, and democratic participation. But the same engineering marvels are shattering centuries-old assumptions about privacy, identity, free expression, and personal control as more and more details of our lives are captured as digital data. Can you control who sees all that personal information about you? Can email be truly confidential, when nothing seems to be private? Shouldn't the Internet be censored the way radio and TV are? is it really a federal crime to download music? When you use Google or Yahoo! to search for something, how do they decide which sites to show you? Do you still have free speech in the digital world? Do you have a voice in shaping government or corporate policies about any of this? Blown to Bits offers provocative answers to these questions and tells intriguing real-life stories. This book is a wake-up call To The human consequences of the digital explosion.

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