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1. An empty 4.00 L steel vessel is filled with 1.00 atm of CH_4 (g) and 4.00 atm of O_2 (g) at 300 °C.

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a mole, and is We use the mole to count atoms and molecules because otherwise, we would have to write out very large numbers to show how many atoms were in...

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sulfur 4 Calculate the number of moles in each of the following quantities a 635' g lithium o q I s • Mol b 346 g zinc S2q mol c 115 g nickel 5

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Supplemental Problems Chemistry: Matter and Change • Chapter 2 1 Data Analysis

Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL.

Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

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Answer Keys - HONORS CHEMISTRY

Supplemental Problems for Chapter 10 1) One overarching trend in atomic structures is that as the coordination number increases,

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the interatomic separation increases. Using the appropriate equations, answer the following questions. (i) Can this trend be demonstrated using the 6-12 potential for van der Waals bonding,

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